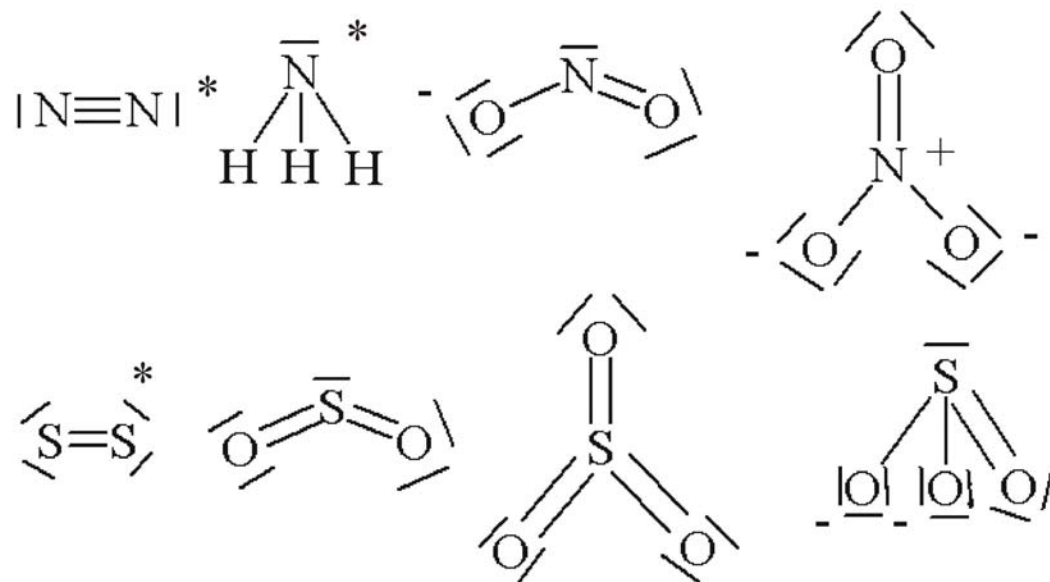


1. Write a Lewis structure for N_2 , NH_3 , NO_2^- , NO_3^- , S_2 , SO_2 , SO_3 , SO_3^{2-} and mark those which are not necessarily resonance structures/hybrids.
 → Octett rule, hypervalence, formal charges



* No other resonance structure