Structure building properties of lone electron pairs

Structural/stereochemical lone pair requirements/activities can lead to different polymorphs as e.g. for CoSeO₃·H₂O

New polymorphic and pseudosymmetrical cobalt selenite monohydrates, CoSeO₃ · H₂O II and CoSeO₃ · H₂O III

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Dedicated to Professor W. H. E. Schwarz on the occasion of his 60th birthday

Abstract. The crystal structures of two polymorphic, hydrothermally synthesized cobalt selenite monohydrates, CoSeO₃ · H₂O II (a = 568.1(1) pm, b = 476.7(1) pm,c = 1348.0(2) pm, $\beta = 101.37(1)^{\circ}$) and CoSeO₃ · H₂O III (a = 477.2(1) pm, b = 1320.9(3) pm, c = 568.6(1) pm, $\beta = 90.54(1)^{\circ}$), both mP32, P2₁/n, Z = 4, have been determined by single crystal X-ray diffraction. Both structures are built up from buckled trans layers $^2_{\infty}$ [CoSeO₃ · H₂O] consisting of strongly distorted CoO₅(H₂O) octahedra and trigonal pyramidal SeO₃²⁻ ions. The structures differ with respect to the stacking of the layers, which originates in their pseudosymmetry and in the stereochemical requirements of the selenite lone electron pair. Only intra-layer H bond schemes are built, i.e., the layers are held together by van der Waals interactions.

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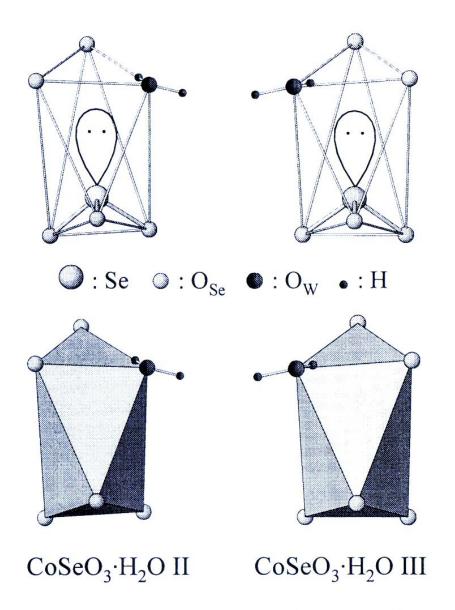


Fig. 5. Environments of the Se atoms between the $_{\infty}^2$ [CoSeO₃ · H₂O] layers of CoSeO₃ · H₂O II (viewed along [010]) and CoSeO₃ · H₂O III (viewed along [100]).

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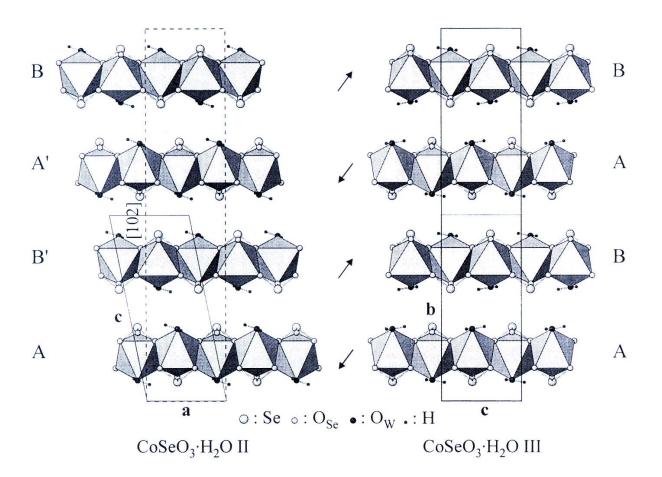


Fig. 4. Projections of the crystal structures of $CoSeO_3 \cdot H_2O$ II (along [010]) and $CoSeO_3 \cdot H_2O$ III (along [100]), showing the stacking of the layers $_{\infty}^2[CoSeO_3 \cdot H_2O]$. Dashed lines belong to the pseudo-B centered cell of CoII. The arrows indicate the directions of the apical Se—O bonds in the respective layers.